

Zhong-Lu You

Department of Chemistry and Chemical
Engineering, Liaoning Normal University,
Dalian 116029, People's Republic of ChinaCorrespondence e-mail:
youzhonglu@yahoo.com.cn

Key indicators

Single-crystal X-ray study
 $T = 298\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.005\text{ \AA}$
 R factor = 0.055
 wR factor = 0.120
Data-to-parameter ratio = 16.1For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

Bis[2-(cyclopropyliminomethyl)-4-nitrophenolato]zinc(II)

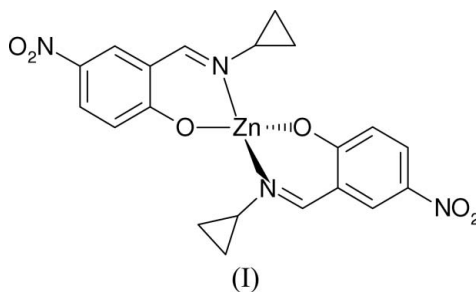
In the title compound, $[\text{Zn}(\text{C}_{10}\text{H}_9\text{N}_2\text{O}_3)_2]$, the Zn^{II} centre displays a distorted tetrahedral configuration defined by two imine N and two phenolate O atoms from two Schiff base ligands.

Received 25 October 2005

Accepted 31 October 2005

Online 5 November 2005

Comment

The crystal structures of a few Schiff base zinc(II) compounds have been previously reported (You, 2005*a,b,c*). As an extension of the work on these compounds, the title zinc(II) compound, (I), is reported here.

The bond lengths and angles in (I), a mononuclear zinc(II) compound (Fig. 1), are comparable to those reported for bis[2-(cyclopropyliminomethyl)phenolato]zinc(II) (You *et al.*, 2003). The central Zn^{II} atom is four-coordinate and is bound by two imine N and two phenolate O atoms from two Schiff base ligands. This ZnN_2O_2 centre has a distorted tetrahedral geometry, with angles subtended at the Zn^{II} atom in the range $96.65(9)$ – $118.69(10)^\circ$ (Table 1). In the crystal structure, weak $\text{C}-\text{H}\cdots\text{O}$ interactions are observed (Table 2).

Experimental

Cyclopropylamine and 5-nitrosalicylaldehyde were available commercially and were used without further purification. Cyclo-

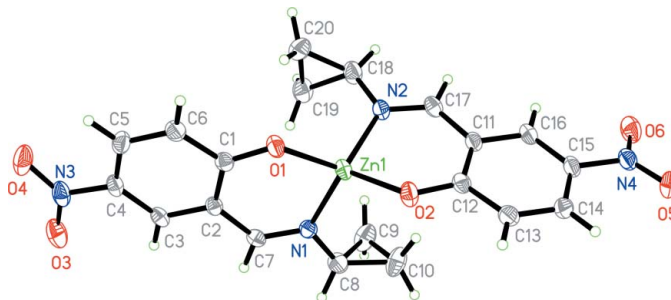


Figure 1

The structure of (I), showing the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level.

propylamine (0.1 mmol, 5.7 mg) and 5-nitrosalicylaldehyde (0.1 mmol, 16.7 mg) were dissolved in MeOH (10 ml). The mixture was stirred at room temperature for 10 min to give a clear yellow solution. To this solution was added a MeOH solution (5 ml) of $\text{Zn}(\text{CH}_3\text{COO})_2 \cdot 4\text{H}_2\text{O}$ (0.1 mmol, 25.6 mg), with stirring. The resulting mixture was stirred for another 10 min at room temperature. After keeping the filtrate in air for 3 d, colourless plate-shaped crystals were formed at the bottom of the vessel.

Crystal data

$[\text{Zn}(\text{C}_{10}\text{H}_9\text{N}_2\text{O}_3)_2]$	$D_x = 1.593 \text{ Mg m}^{-3}$
$M_r = 475.75$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 3240 reflections
$a = 11.363 (1) \text{ \AA}$	$\theta = 2.3\text{--}24.2^\circ$
$b = 16.873 (2) \text{ \AA}$	$\mu = 1.29 \text{ mm}^{-1}$
$c = 11.266 (1) \text{ \AA}$	$T = 298 (2) \text{ K}$
$\beta = 113.283 (1)^\circ$	Plate, colourless
$V = 1984.1 (3) \text{ \AA}^3$	$0.18 \times 0.13 \times 0.07 \text{ mm}$
$Z = 4$	

Data collection

Bruker SMART CCD area-detector diffractometer	4519 independent reflections
ω scans	3579 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	$R_{\text{int}} = 0.048$
$T_{\text{min}} = 0.802$, $T_{\text{max}} = 0.915$	$\theta_{\text{max}} = 27.5^\circ$
16852 measured reflections	$h = -14 \rightarrow 14$
	$k = -21 \rightarrow 21$
	$l = -14 \rightarrow 14$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0499P)^2 + 0.5393P]$
$R[F^2 > 2\sigma(F^2)] = 0.055$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.120$	$(\Delta/\sigma)_{\text{max}} = 0.001$
$S = 1.08$	$\Delta\rho_{\text{max}} = 0.46 \text{ e \AA}^{-3}$
4519 reflections	$\Delta\rho_{\text{min}} = -0.32 \text{ e \AA}^{-3}$
280 parameters	
H-atom parameters constrained	

Table 1

Selected geometric parameters (\AA , $^\circ$).

Zn1—O1	1.908 (2)	Zn1—N2	2.010 (3)
Zn1—O2	1.909 (2)	Zn1—N1	2.019 (3)
O1—Zn1—O2	118.69 (10)	O1—Zn1—N1	96.65 (9)
O1—Zn1—N2	115.15 (10)	O2—Zn1—N1	114.47 (11)
O2—Zn1—N2	96.93 (10)	N2—Zn1—N1	116.33 (10)

Table 2

Hydrogen-bond geometry (\AA , $^\circ$).

$D\text{---}H\cdots A$	$D\text{---}H$	$H\cdots A$	$D\cdots A$	$D\text{---}H\cdots A$
C13—H13 \cdots O4 ⁱ	0.93	2.59	3.366 (4)	142
C17—H17 \cdots O3 ⁱⁱ	0.93	2.44	3.349 (5)	165
C20—H20A \cdots O1	0.97	2.58	3.398 (4)	142

Symmetry codes: (i) $-x + 1, -y + 2, -z + 2$; (ii) $x - 1, y, z - 1$.

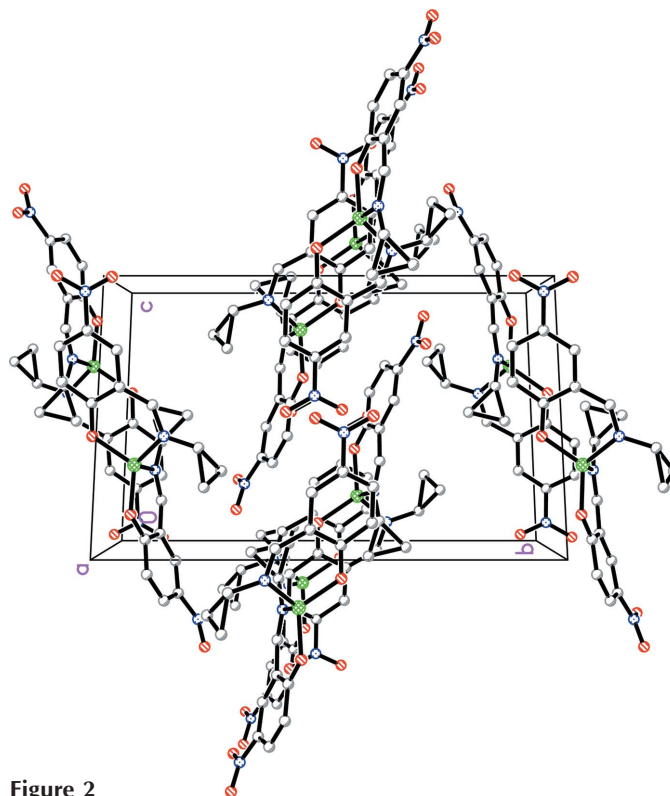


Figure 2

The crystal packing of (I), viewed along the a axis. H atoms have been omitted.

H atoms were placed in idealized positions and constrained to ride on their parent atoms, with $\text{C—H} = 0.93\text{--}0.98 \text{ \AA}$ and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1998); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997a); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997a); molecular graphics: SHELXTL (Sheldrick, 1997b); software used to prepare material for publication: SHELXTL.

The author thanks the Liaoning Normal University, People's Republic of China, for funding this study.

References

- Bruker (1998). SMART (Version 5.628) and SAINT (Version 6.02). Bruker AXS Inc., Madison, Wisconsin, USA.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997a). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Sheldrick, G. M. (1997b). SHELXTL. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.
- You, Z.-L. (2005a). Acta Cryst. E61, m1571–m1573.
- You, Z.-L. (2005b). Acta Cryst. C61, m456–m458.
- You, Z.-L. (2005c). Acta Cryst. C61, m383–m385.
- You, Z.-L., Lin, Y.-S., Liu, W.-S., Tan, M.-Y. & Zhu, H.-L. (2003). Acta Cryst. E59, m1025–m1027.